

T.P. n° 2 : Determination of the initial speed of an enzymatic reaction

Principle :

To measure the initial speed, the experiment is carried out in the presence of enzyme and substrate at constant concentration, in a buffered medium with a pH of 4.7 and thermostated at 37°C. The contact time between the substrate and enzyme is varied and the concentration of the product formed (inverted sugars) is measured.

Equipment :

- ✓ Test tubes and rack ;
- ✓ Pipettes and micropipettes ;
- ✓ Pipettors ;
- ✓ Beakers ;
- ✓ Balance ;
- ✓ Watch glass and spatula ;
- ✓ Agitator and magnetic stir bar ;
- ✓ Oven set to 37°C ;
- ✓ Boiling water bath ;
- ✓ Vortex agitator ;
- ✓ Spectrophotometer and cuvettes.

Reagents :

- ✓ Diluted enzyme extract (1/50) ;
- ✓ 0.05M acetate buffer at pH 4.7 ;
- ✓ M sucrose solution ;
- ✓ DNS reagent.

Procedure :

- ✓ Prepare the tubes according to the table below. :

Tube number	0	1	2	3	4	5	6
0.1M sucrose solution (ml)	1						
Distilled water (ml)	1						
Acetate buffer pH 4.7 (ml)	1						
Preincubation (min)	5 min at 37 °C						
Diluted enzyme extract (1/50) (ml)	0	0,1					

Contact time (min)	0	1	2	4	6	10	15
DNS reagent (ml)	2						
Incubation	Homogenize, seal the tubes with aluminum foil, and heat in a boiling water bath for 5 minutes . Allow to cool, then add :						
Distilled water (ml)	6						

- ✓ Homogenize and let stand for 10 min at laboratory temperature.
- ✓ Read the absorbances (OD) at 540 nm against the blank (tube 0).

Work to be performed :

- ✓ Trace the curve $DO = f(t)$
- ✓ Trace the curve $[\text{invert sugar}] = f(t)$
- ✓ Calculate the initial reaction speed (V_i) in $\mu\text{mol of hydrolyzed sucrose} \cdot \text{L}^{-1} \cdot \text{min}^{-1}$ (L = liter of reaction medium)).