

## Notions of Transport Phenomena

# Final Exam - correction

### Closed Notes, Show All Work

#### Exercise 01 (08 Marks): “About heat transfer”

- 1- There are: **THREE (3)** main ways of heat transfer. 0,25
  - Conduction HT; 0,25
  - Convection HT, and; 0,25
  - Radiation. 0,25
- 2- Fourier’s law. 1

3-  $q_x'' = -k \frac{dT}{dx}$  (heat flux) or,  $q_x = -kA \frac{dT}{dx}$  (heat transfer rate) 1

The thermal conductivity of most solids varies with temperature. In general:

$$k = \frac{\dot{q}_x}{\partial T / \partial x} \quad 1$$

Most materials are very nearly homogeneous, therefore we can usually write  $k = k(T)$ .  
 As temperature increases, both the number of free electrons and lattice vibrations increase. However, the thermal conductivity of metals decreases slightly with increasing temperature.

#### Part 01:

4-  $q_x'' = h \cdot (T_s - T_{surr})$  (heat flux) or,  $q_x = h \cdot A \cdot (T_s - T_{surr})$  (heat transfer rate) 1

5-  $[h] : \text{W/m}^2 \cdot \text{K}$  1

6-  $q_{conduction}'' = q_{convection}''$  then,  $-k \frac{dT}{dx} = h \cdot (T_s - T_{surr})$  so,  $T_s = -k \frac{dT}{dx} + T_{surr} \dots$  1

#### Part 02:

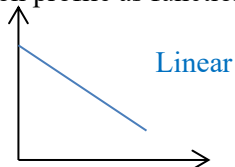
$q_{conduction}'' = q_{radiation}''$  Then,  $-k \frac{dT}{dx} = \sigma \cdot (T_s^4 - T_{surr}^4)$  so... 1

#### Exercise 02 (07 Marks): “About mass transfer”

1- Gradient of concentration :  $\frac{dc}{dx}$  1

2-  $J_{AB} = -D_{AB} \cdot \frac{dc}{dx}$  ; called: Fick's first law of diffusion 2

3- Concentration profile as function of (x):



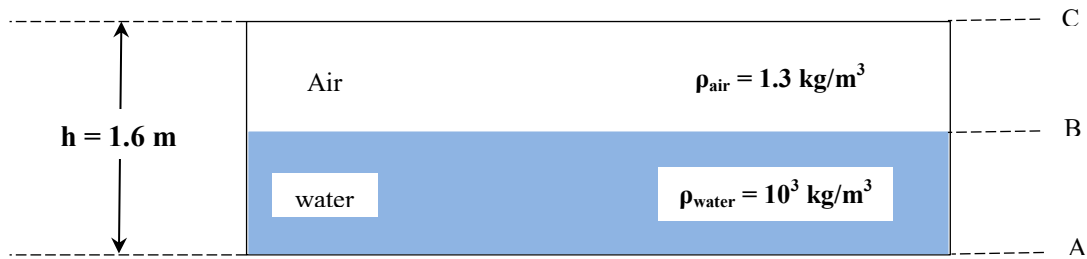
4- The negative sign of the equation indicates that diffusion occurs in a direction opposite to that of the increasing concentration. Hence, diffusion occurs in the direction of decreasing concentration of the diffusing substance, and thus, the diffusion flux is a positive quantity. 1

5- If we change the species (A) and (B) by two gases, the new rate of diffusional mass transfer increase. 1

6- There is two main types of mass transfer, the first one is molecular diffusion, and the second one is: convection mass transfer. 1

**Exercise 03 (05 Marks): “About momentum transfer”**

The tank shown in **figure 03** is half filled water.



- 1- Calculate the pressure difference between points **A** and **B**.

$$\Delta P_{AB} = \rho_{water} \cdot g \cdot h \quad \text{then, } \Delta P_{AB} = 10^3 \cdot 9,81 \cdot \left(\frac{1,6}{2}\right) \quad , \text{ so } \Delta P_{AB} = 7\,848 \text{ Pa} \quad 0,25 \times 4$$

- 2- Calculate the pressure difference between points **B** and **C**.

$$\Delta P_{BC} = \rho_{air} \cdot g \cdot h \quad \text{then, } \Delta P_{BC} = 1,3 \cdot 9,81 \cdot \left(\frac{1,6}{2}\right) \quad , \text{ so } \Delta P_{BC} = 10,2 \text{ Pa} \quad 0,25 \times 4$$

- 3- Compare these results.

$$\Delta P_{AB} \gg \Delta P_{BC} \quad \text{(The pressure differential in water is much greater than in air.)}$$

This is due to the volumetric mass of matter.

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*Good Luck,*  
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