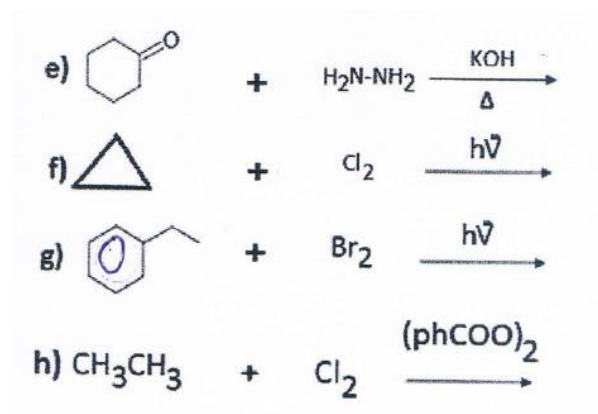
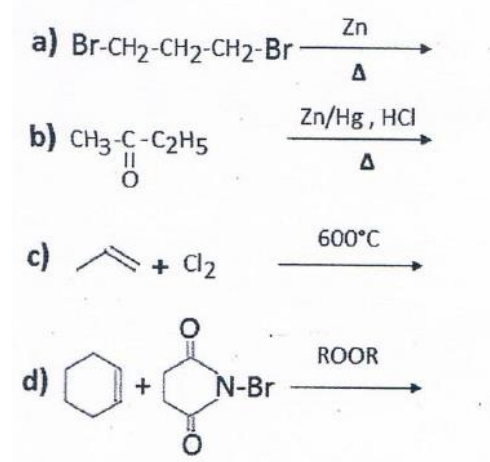


Series N° 2

Exercise 1

complete the following reactions



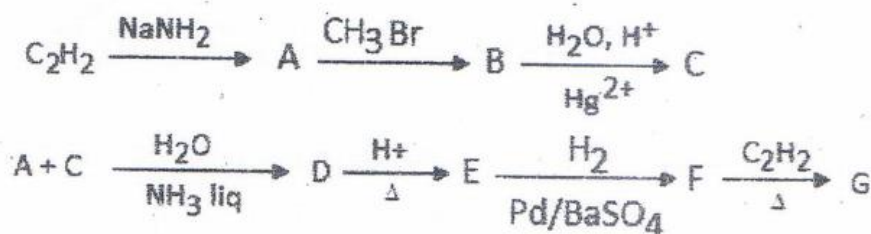
Exercise 2

Three alkanes, A, B, and C, have the same molar mass of 72 g/mol.

- Indicate the possible structures for A, B, and C.
- Assign each structure to A, B, or C, knowing that:
 - A produce three monochlorinated derivatives (D, E, F).
 - B produces four monochlorinated derivatives (G, H, I, J).
 - C produces only one monochlorinated derivative (K) when reacting with chlorine at 300°C.
- write the mechanism of $C \rightarrow K$.
- Determine the structures of D, G, and H, knowing that they possess an asymmetric carbon and that G is the least abundant.

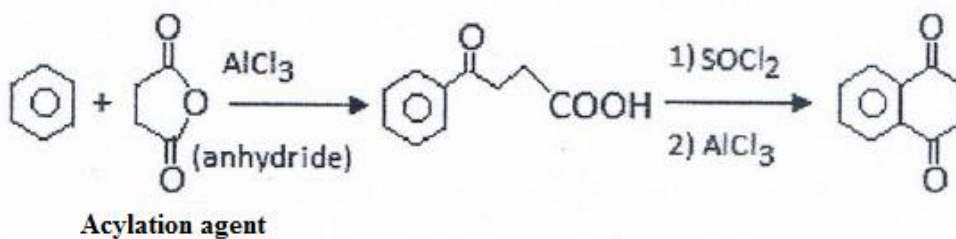
Exercise 3

Identify the compounds a, b, c.....

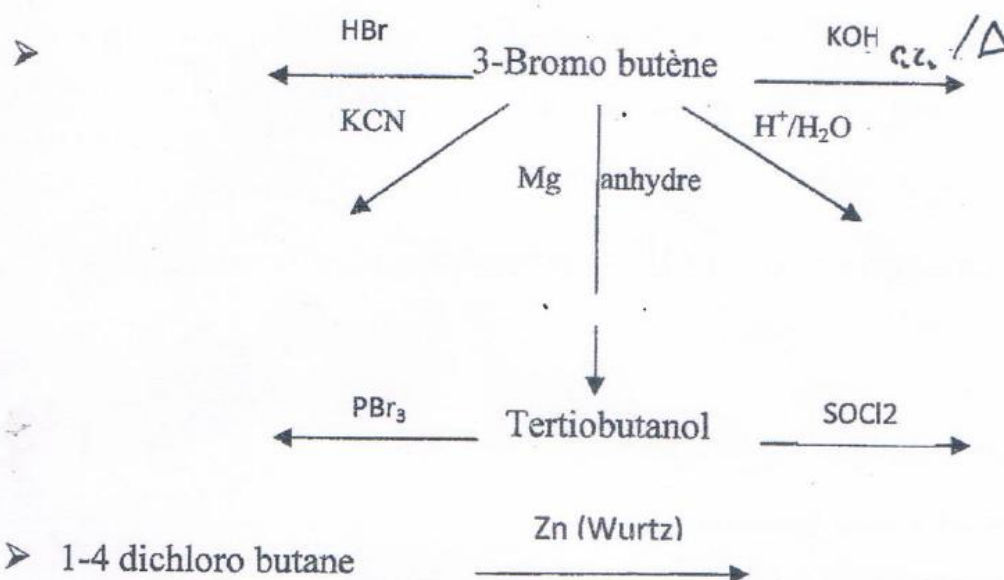


Exercise 4

Detailed the mechanism of the following reaction



Exercise 5



Exercise 6

How to prepare the compound (A) C_2H_5MgBr starting from acetylene.

- What precaution should be taken to preserve it?
- The compound (A) is reacted with ethanal. After hydrolysis, a compound (B) is formed, which upon dehydration gives two ethylenic hydrocarbons (C) and (D).
 - a) Write the different reactions, providing the name(s) of the product(s) obtained.
 - b) Show that the addition of **HBr** to (C) and (D) leads to the same compound.