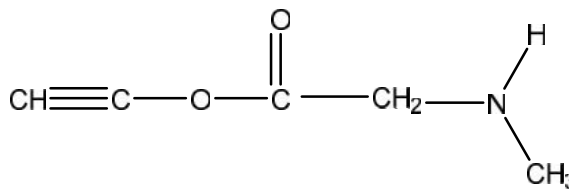
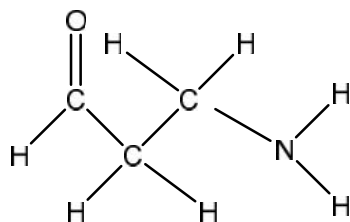
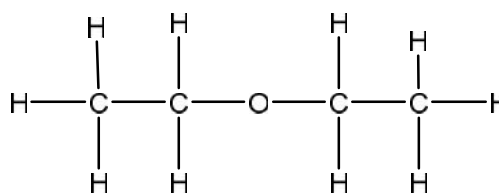
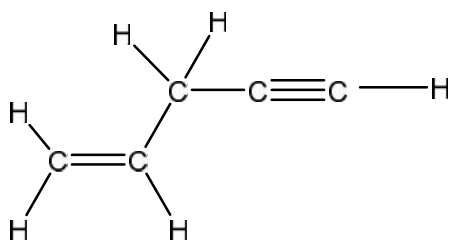




Series N° 01

Exercise 1:

Indicate the hybridization state of C, the  $\delta$  bond number in the following compounds:

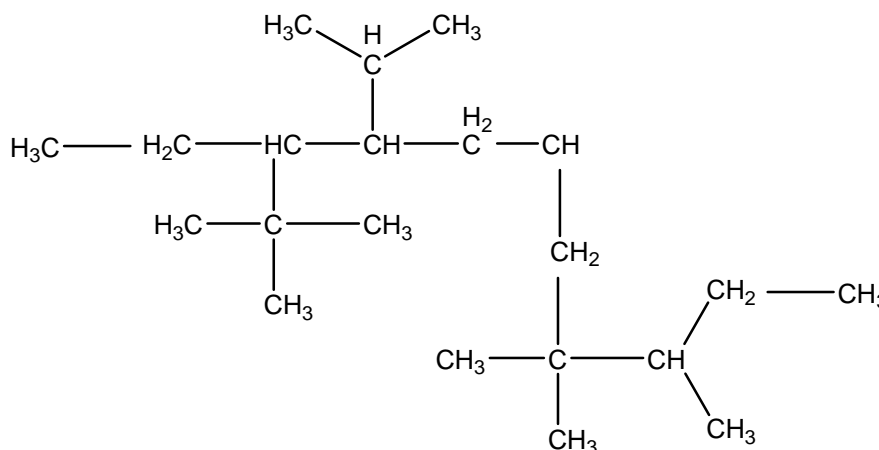


Exercise 2:

Find the number of unsaturations possessed by the compounds corresponding to the following crude formulas:  $C_4H_{10}$ ,  $C_3H_4Cl_2$ ,  $C_3H_9N$ ,  $C_3H_6O$ .

Exercise 3:

Look at the following saturated hydrocarbon:



- Indicate the different types of carbon (primary, secondary, tertiary or quaternary).
- write the crude and topological formula of this compound

2/ represent a chain of seven carbons: Linear, Branched, Cyclical, Polycyclic (bi- and tricycle).



### Exercise 4:

Represent the topological structure of the following compounds and correct the name if necessary

- A. 4-isopropyl-5-propyloctane
- B. Butan-2,3-dione
- C. Butane-1,4-dial
- D. Hexan-2-one
- E. Ethanoic acid
- F. Cyclo-2-pentanol
- G. Butadiene
- H. 4-aminobenzoic acid
- I. N-butylpropane
- J. Cyclopent-1-ene-1-ol
- K. Methyl ethanoat
- L. 8-methylbicyclo [4.2.1] non-6-ene

### Exercise 5:

Give the name according to the I.U.P.A.C nomenclature of the following compounds:

