

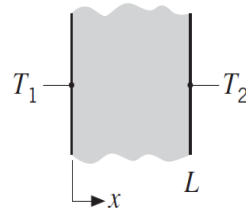
Tutorial Two

| Introduction to conduction |

Exercise 2.1

Fourier’s Law

Consider steady-state conditions for one-dimensional conduction in a plane wall having a thermal conductivity $k = 50 \text{ W/m.K}$ and a thickness $L = 0.35 \text{ m}$, with no internal heat generation.

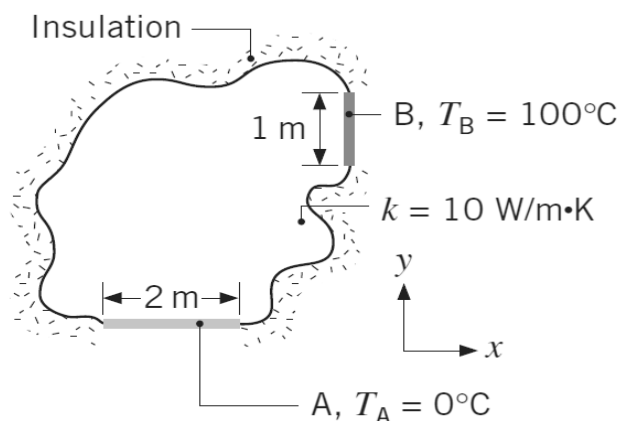


Determine the heat flux and the unknown quantity for each case and sketch the temperature distribution, indicating the direction of the heat flux.

Case	$T_1(^{\circ}\text{C})$	$T_2(^{\circ}\text{C})$	$dT/dx \text{ (K/m)}$
1	50	-20	
2	-30	-10	
3	70		160
4		40	-80
5		30	200

Exercise 2.2

In the two-dimensional body illustrated, the gradient at surface **A** is found to be $\partial T/\partial y = 30 \text{ K/m}$. What are $\partial T/\partial y$ and $\partial T/\partial x$ at surface **B**?

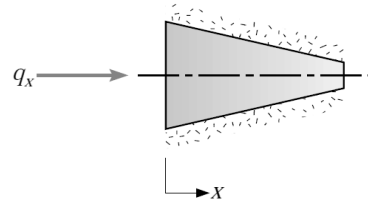


For the case where the thermal conductivity varies with temperature as $k = k_0 + aT$, where $k_0 = 10 \text{ W/m.K}$, $a = -10^{-3} \text{ W/m.K}^2$, and T is in kelvins. The gradient at surface **B** is $\partial T/\partial x = 30 \text{ K/m}$. What is $\partial T/\partial y$ at surface **A**?

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Exercise 2.3

Assume steady-state, one-dimensional heat conduction through the symmetric shape shown.



Assuming that there is no internal heat generation, derive an expression for the thermal conductivity $k(x)$ for these conditions: $A(x) = (1 - x)$, $T(x) = 300(1 - 2x - x^3)$, and $q = 6000 \text{ W}$, where A is in square meters, T in kelvins, and x in meters.

Exercise 2.4 Thermophysical Properties

Consider a **300 mm x 300 mm** window in an aircraft. For a temperature difference of **80 °C** from the inner to the outer surface of the window, calculate the heat loss through **L = 10-mm-thick** polycarbonate, soda lime glass, and aerogel windows, respectively.

Material	Thermal conductivity	unit
Polycarbonate (k_{pc})	0,21	(W/m.K)
Soda lime glass (k_{sig})	1,4	
Aerogel (k_{ag})	0,014	

If the aircraft has **130 windows** and the cost to heat the cabin air is **£1/kW.h**, compare the costs associated with the heat loss through the windows for an **8-hour** intercontinental flight.

Exercise 2.5

Test your understanding of various concepts by addressing the following questions:

- 1- An important property of matter is defined by *Fourier's law*. What is it? What is its physical significance? What are its units?
- 2- What is an *isotropic* material?
- 3- Why is the thermal conductivity of a solid generally larger than that of a liquid? Why is the thermal conductivity of a liquid larger than that of a gas?